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precipitate silver iodide. chemical equation: $KI(aq) + AgNO_3(aq) \rightarrow KNO_3(aq) + AgI(s)$ complete ionic equation: $K^+(aq) + I^-(aq) + Ag^+(aq) + NO_3^-(aq) \rightarrow K^+(aq) + NO_3^-(aq) + AgI(s)$

Chemical Reactions

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CHAPTER 6 SOLUTIONS MANUAL 4. Determine why producing 1 million francium atoms per second is not enough to make measurements, such as density of boiling point. Even one million atoms collected together as a solid are microscopic. A grain of salt contains about 10^{15} sodium atoms.

The Periodic Table and Periodic Law

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45. O_3 , O_2 , O , O_4 , O_6 , O_8 , O_{10} , O_{12} , O_{14} , O_{16} , O_{18} , O_{20} , O_{22} , O_{24} , O_{26} , O_{28} , O_{30} , O_{32} , O_{34} , O_{36} , O_{38} , O_{40} , O_{42} , O_{44} , O_{46} , O_{48} , O_{50} , O_{52} , O_{54} , O_{56} , O_{58} , O_{60} , O_{62} , O_{64} , O_{66} , O_{68} , O_{70} , O_{72} , O_{74} , O_{76} , O_{78} , O_{80} , O_{82} , O_{84} , O_{86} , O_{88} , O_{90} , O_{92} , O_{94} , O_{96} , O_{98} , O_{100} .
46. Challenge Draw the Lewis resonance structure for the ion SO_3^{2-} . $O=S(O)_2-O^-$
47. Draw the expanded octet Lewis structure of each molecule.
48. PCl_5 , Cl_2 , Cl_4 , Cl_6 , Cl_8 , Cl_{10} , Cl_{12} , Cl_{14} , Cl_{16} , Cl_{18} , Cl_{20} , Cl_{22} , Cl_{24} , Cl_{26} , Cl_{28} , Cl_{30} , Cl_{32} , Cl_{34} , Cl_{36} , Cl_{38} , Cl_{40} , Cl_{42} , Cl_{44} , Cl_{46} , Cl_{48} , Cl_{50} , Cl_{52} , Cl_{54} , Cl_{56} , Cl_{58} , Cl_{60} , Cl_{62} , Cl_{64} , Cl_{66} , Cl_{68} , Cl_{70} , Cl_{72} , Cl_{74} , Cl_{76} , Cl_{78} , Cl_{80} , Cl_{82} , Cl_{84} , Cl_{86} , Cl_{88} , Cl_{90} , Cl_{92} , Cl_{94} , Cl_{96} , Cl_{98} , Cl_{100} .
49. Challenge Draw the Lewis structure for the molecule formed when six fluorine atoms and

Covalent Bonding

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