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Chemistry Unit 3 Test Answer

Unit 3 chemistry test. - arranged the known elements in order of increasing atomic mass -noticed a repeating pattern in the properties of the elements -designed a table with rows and columns to show the repeating pattern -left blank spaces in the

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table to represent elements not known in his time -predicted the chemical properties of those elements.

Unit 3 Chemistry Test Answers

Q. A student collected this information when performing the Combustion of a Candle experiment: $E = M \times C \times \Delta T$. Volume of Water = 104 mL. Water Temperature = 17°C. Specific Heat of

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Water = $4.18 \text{ J/g}^\circ\text{C}$.
Calculate the heat (J)
absorbed by the water
using the data
provided.

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1. that some substances can dissolve in solvents.
2. heat added to a substance.
3. random internal motion of atoms and molecules.
- 4, attraction of

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opposite atomic charges. Cooling reduces particle motion, resulting in coalescence by attractive forces.

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Chemistry Unit 3 Test
Answers 1. B 2. A 3. C
4. A 5. C 6. C 7. A 8. C
9. B 10. C 11. B 12. D
13. A 14. B 15. A 16.

Three of the following:

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- a. All matter is composed of indivisible particles called atoms
- b. All atoms of an element are identical, atoms of different elements have different properties.

Chemistry Unit 3
Exam Answers
Quantitative Energy
Problems

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PRACTICE TEST Use

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these questions as a guide as a START to your studying. Each question below represents a concept and type of question that you should feel confident completing. If you have difficulty with one of the questions below, then you should go back to

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CHEMISTRY UNIT TEST
-1 KEY ANSWER PART -

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I Each question ONE

Mark $14 \times 1 = 14$

Q.No. Answer Q.No.

Answer 1. b 8. c 2.

Number of Csa 9. c 3. c
unit cell is one 10. a 4.

a 11. d 5. c 12. c 6. c

13. d 14. d

PART-II Answer any

Seven of the following

Q. No. Content Mark

Total 15. BeCl_2 Total

valence electrons = 2

+ $7 \times 2 = 16$ Examples

...

N 3 Answer any

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Seven of the following Content Ma rk

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vocabulary, and
conventions are used
correctly. Some
science symbols=
vocabulary,

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Key - Most science
symbols vocabulary**

...

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Fatal error: Uncaught
Error: Call to undefined
function

mysql_connect() in /ho
me/findtest/include/con
nect.php:6 Stack trace:
#0 /home/findtest/publ
ic_html/index.php(2 ...

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Chemistry (H): Unit 3 -

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The Atom - Test
Multiple Choice Identify the choice that best completes the statement or answers the question.

1. The number of neutrons in a "normal" potassium atom is

a. 125 d. 20 b. 16 e. none of the answers are correct c. 15

2. The atomic mass of manganese is

a. 25 c. 12 b. 54.94 d. 24.31

3.

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Unit 1 Test (Nov 2003)
Test ID A: Question 3
Test ID B: Question 7

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Ionic Bonding: 1. Ionic bond is the attraction between oppositely charged ions (electrostatic force of attraction). 2. Charged ions are created when valence electrons are removed (cation / positively charged ions) or electrons are added to the

Multiple Choice
Answers Test ID B 1.
D 1. C 2. D 2. C 3. C

...

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Unit 3: Chemistry -
Matter and Elements
Unit 3 Review Packet
TEKs: 6.5A - know that
an element is a pure
substance represented
by chemical symbols.
6.5B - recognize that a
limited number of the
many known elements
comprise the largest
portion of solid Earth,
living matter, oceans,
and the atmosphere.

Unit 3: Chemistry
Matter and Elements

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Chemistry Unit 3 Test
Answers 1. B 2. A 3. C
4. A 5. C 6. C 7. A 8. C
9. B 10. C 11. B 12. D
13. A 14. B 15. A 16.

Three of the following:

- a. All matter is composed of indivisible particles called atoms
- b. All atoms of an element are identical, atoms of different elements have different properties

C.

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1s²2s²2p⁶3s²3p⁶4s² c

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chemical phenomena.

... Unit 3:

Intermolecular Forces
and Properties

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